

Welcome to ChemThink



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Signup Students

Your code is given below. Your students should use this code to register the first time they visit the site. This code is valid for 14 days. If you need another code after that time, simply return here and request another.

code: 9600 - 2780 - 5564

expires: 1/22/2015

If you would like you may print out a page of instructions that will lead your students through the registration process by clicking the button below.

[Print Instructions](#)

The following instructions will guide you through the process of registering for ChemThink.

1. Open a browser and go to the ChemThink site at www.chemthink.com
2. On the right hand side of the screen, click on the words 'Not Registered?'
3. Enter the code written below to identify you as one of your teacher's students. Then click on the Register button.

Your code is: 9600 - 2780 - 5564

4. This will call up the registration screen where you will enter your first name, last name, a user name, and password. You must also choose the period for your class.
5. Click on the Register Now button to complete the process.

Notes:

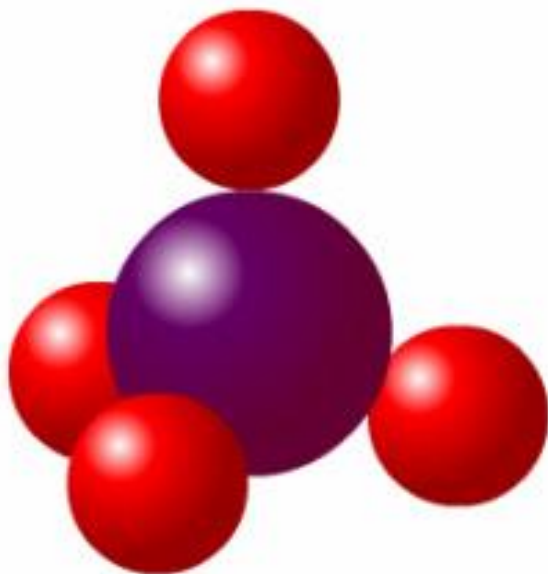
You will only need to register once. After registering you will enter the site using the username and password you have chosen.

Choose a user name and password that you will remember.

The code given above is good until 1/22/2015, after that you will have to ask your teacher for a new code.

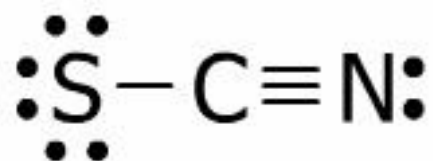
"In this tutorial we will investigate the shapes of simple covalent molecules. Specifically, we will be learning about something called VSEPR theory and how it allows us to predict the shapes that covalently bonded molecules will have."

The molecular structure shown below would be best described by which of the following names?

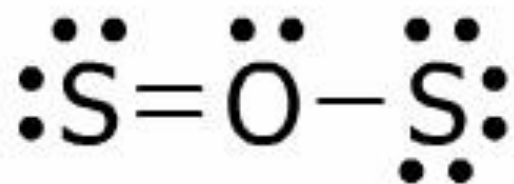


- linear
- trigonal planar
- bent
- tetrahedral
- trigonal pyramid

How many areas of electron concentration are found around the central atom in the molecule below?



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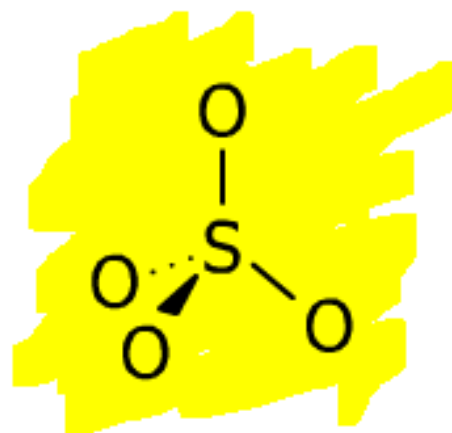
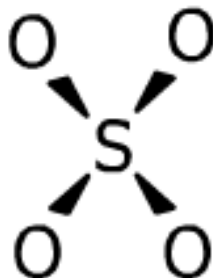
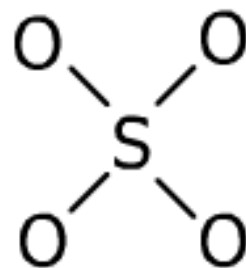
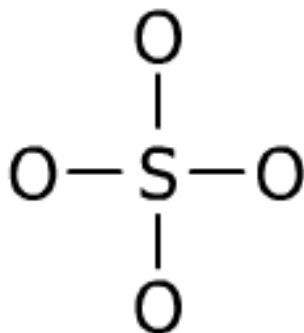


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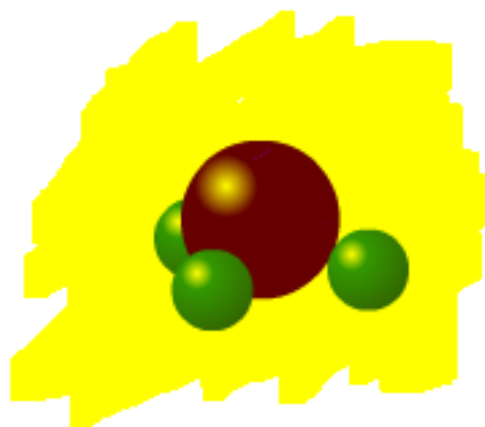
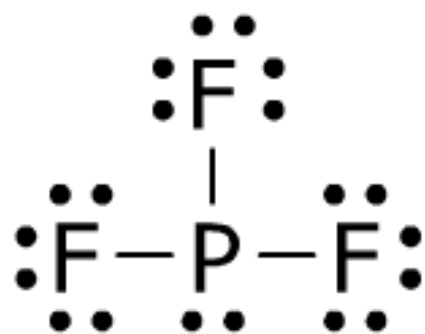


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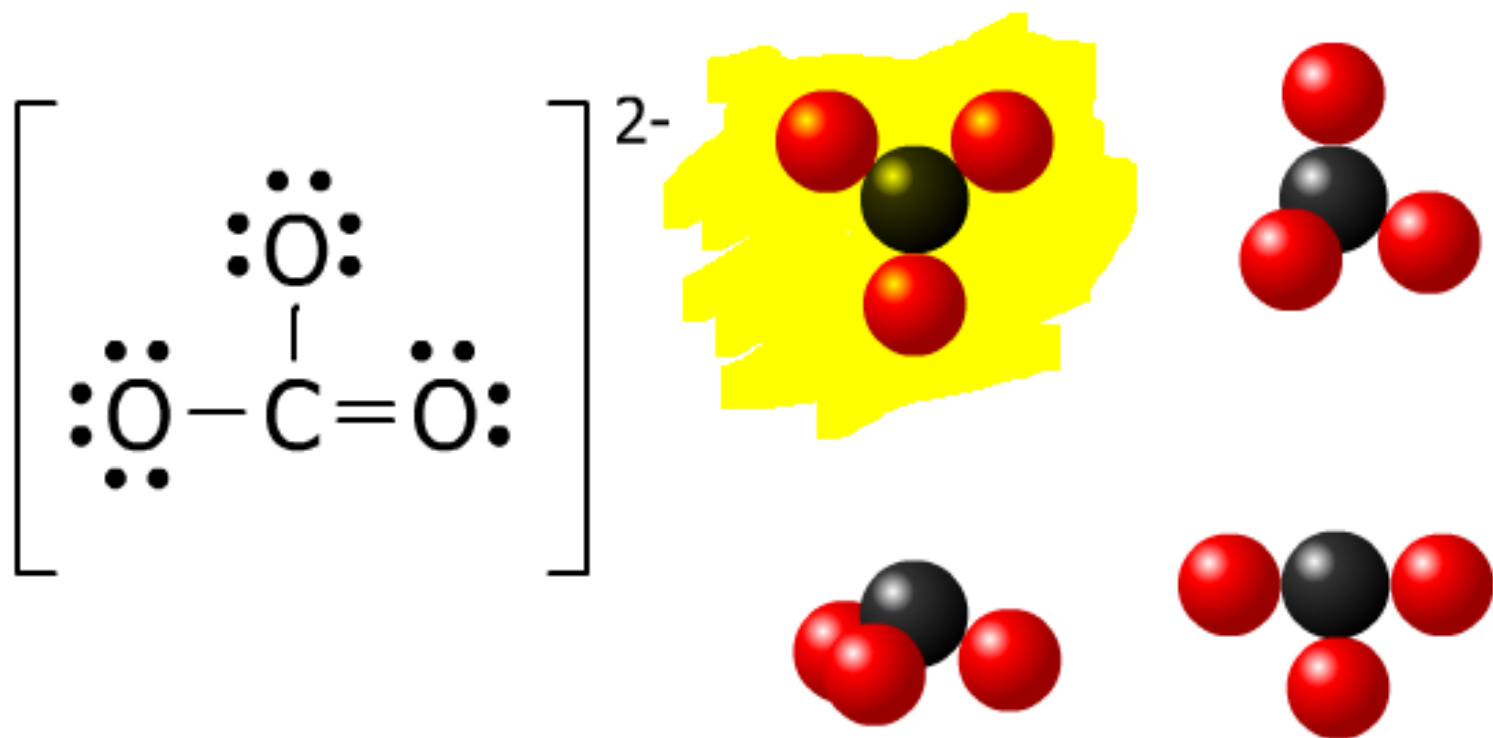
Click on the two dimensional drawing below that best represents the pictured molecule.



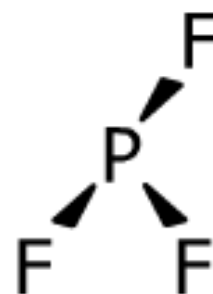
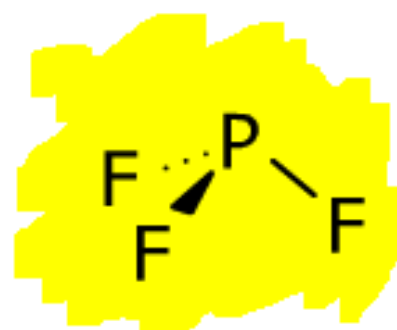
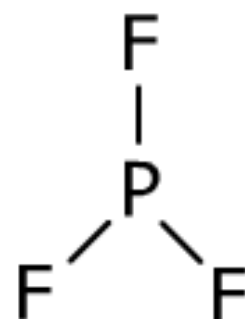
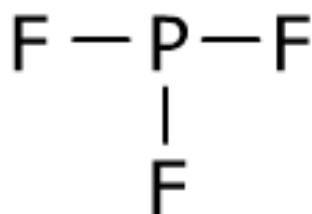
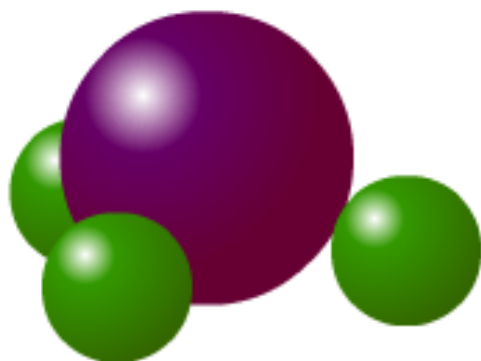
Click on the molecule that best represents the correct shape for the following Lewis structure.



Click on the molecule that best represents the correct shape for the following Lewis structure.



Click on the two dimensional drawing below that best represents the pictured molecule.



Which of the following names correctly identifies the structure of the molecule whose formula is given below?



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