## Measurement Review

## ***ALL ANSWERS MUST INCLUDE THE PROPER UNITS AND NUMBER OF SIG FIGS*** CALCULATE PERCENT ERROR FOR THE FOLLOWING VALUES:

1. Marisa determined the melting point of a substance to be $24.5^{\circ} \mathrm{C}$. Find the percent error of her measurement if the actual melting point is $31.2^{\circ} \mathrm{C}$.
2. The molar mass of butane is $58.14 \mathrm{~g} / \mathrm{mol}$. Using his lab data, Tyrone calculated the molar mass of butane as $44.2 \mathrm{~g} / \mathrm{mol}$. Find the percent error of his measurement.

DETERMINE THE NUMBER OF SIGNIFICANT FIGURES IN THE FOLLOWING NUMBERS:
3. $320,000 \mathrm{~mm}$
4. $\quad 0.0400 \mathrm{~g}$
5. $5,000 \mathrm{~km}$
6. $68,050 \mu \mathrm{~L}$

CONVERT THE FOLLOWING NUMBERS INTO OR OUT OF SCIENTIFIC NOTATION:
7. 0.000506 mL
8. $42,000,000,000 \mathrm{pm}$
9. $5.00 \times 10^{-3} \mathrm{~km}$
10. $8.200 \times 10^{2} \mathrm{~m}$

## CALCULATE AND EXPRESS ANSWERS IN THE CORRECT UNITS AND \# OF SIG FIGS.

11. $(0.00600 \mathrm{~m}) \div(0.030 \mathrm{~s})=$
12. $(167.55 \mathrm{~g})-(87.3 \mathrm{~g})=$
13. $(50.75 \mathrm{~mL})+(155 \mathrm{~mL})=$
14. $(5,200 \mathrm{~cm})(0.07 \mathrm{~cm})=$
15. $(12.5 \mathrm{~g}) \div\left(6.0 \mathrm{~g} / \mathrm{cm}^{3}\right)=$
16. $(370 \mathrm{mg})+(1200 \mathrm{mg})=$

SOLVE THE FOLLOWING DENSITY PROBLEMS:
17. Limestone has a density of $2.72 \mathrm{~g} / \mathrm{cm}^{3}$. What is the mass of $24.9 \mathrm{~cm}^{3}$ of limestone?
18. Helium has a density of $0.017 \mathrm{~g} / \mathrm{L}$. What is the volume of a weather balloon that contains 38 g of helium?
19. A $0.750-\mathrm{cm}^{3}$ sample of platinum has a density of $21.4 \mathrm{~g} / \mathrm{cm}^{3}$. What is its mass?

PERFORM THE FOLLOWING SI UNIT CONVERSIONS (watch sig figs!):
20. $177 \mathrm{~mL}=$ $\qquad$ L
22. $0.093 \mathrm{~kg}=$ $\qquad$ mg
21. $56 \mathrm{~m}=$ $\qquad$ cm
23. $54,400 \mu \mathrm{~m}=$ $\qquad$ dm

USE THE FACTOR-LABEL METHOD TO SOLVE THE FOLLOWING PROBLEMS:
24. George walks 1.5 km to school. If each step he takes is equal to 2.25 ft , how many steps does he take?
25. Susanna is 5.50 ft tall. What is her height in centimeters?
26. A can of Diet Pepsi ${ }^{\oplus}$ contains 355 mL of soda. How many cans would have to be opened in order to fill a $1.0-\mathrm{m}^{3}$ tank?
27. How many milliliters are in a 20.0-oz. bottle of soda? (There are 32 oz . in 1 quart.)
28. An ant is about 4.0 mm long. How many ants does it take to span 2.0 feet?
29. One serving of $\mathrm{Jell}^{\circledR}{ }^{\circledR}$ instant pudding requires 28.0 g of mix. If each box contains 107 g of mix, how many boxes are required to serve 15 people?
30. How many pounds does 1.0 quart of motor oil weigh if the density of motor oil is $0.80 \mathrm{~g} / \mathrm{mL}$ ?

## Measurement Review ANSWER KEY

***ALL ANSWERS MUST INCLUDE THE PROPER UNITS AND NUMBER OF SIG FIGS***

1. $21 \%$ or $21.5 \%$
2. $23.9 \%$ or $24.0 \%$
3. 2
4. 3
5. 1
6. 4
7. $5.06 \times 10^{-4} \mathrm{~mL}$
8. $4.2 \times 10^{10} \mathrm{pm}$
9. 0.00500 km
10. 820.0 m
11. $0.20 \mathrm{~m} / \mathrm{s}$
12. 80.3 g
13. 206 mL
14. $400 \mathrm{~cm}^{2}$
15. $2.1 \mathrm{~cm}^{3}$
16. 1600 mg
17. 67.7 g
18. $2,200 \mathrm{~L}$
19. 16.1 g
20. 0.177 L
21. $5,600 \mathrm{~cm}$
22. $93,000 \mathrm{mg}$
23. 0.544 dm
24. 2,200 steps
25. 168 cm
26. 2800 cans
27. 591 mL
28. 15 ants
29. 3.93 boxes
30. 1.7 lbs
